# **JEE Advanced 2026**

# Sample Paper - 1 (Paper-2)

Time Allowed: 3 hours Maximum Marks: 180

**General Instructions:** 

This question paper has THREE main sections and three sub-sections as below.

**MCQ** 

- Each question has FOUR options (A), (B), (C) and (D). ONLY ONE of these four options is the correct answer.
- You will get +3 marks for the correct response and -1 for the incorrect response.

**MRQ** 

- Each question has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is(are) the correct answer(s).
- You will get +4 marks for the correct response and -2 for the incorrect response.
- You will also get 1-3 marks for a partially correct response.

**NUM** 

- The answer to each question is a NON-NEGATIVE INTEGER.
- You will get +4 marks for the correct response and 0 marks for the incorrect response.

**Physics** 

1. Two identical thin rings, each of radius R metres, are coaxially placed a distance R metres apart. If Q<sub>1</sub> coulomb and Q<sub>2</sub> coulomb, are respectively the charges uniformly spread on the two rings, the work done in moving a charge q from the centre of one ring to that of the other is:

a) 
$$\frac{q(Q_1 - Q_2)(\sqrt{2} - 1)}{(4\sqrt{2}\pi\varepsilon_0 R)}$$

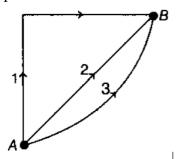
b) 
$$\frac{q(Q_1+Q_2)(\sqrt{2}+1)}{(4\sqrt{2}\pi\varepsilon_0R)}$$

d) 
$$\frac{q\sqrt{2}(Q_1+Q_2)}{(4\pi\varepsilon_0 R)}$$

- 2. The ratio of the magnetic field and magnetic moment at the centre of a current-carrying circular loop is x. When both the current and radius is doubled the ratio will be:
  - a)  $\frac{x}{8}$

b)2x

3. If W<sub>1</sub>,W<sub>2</sub> and W<sub>3</sub> represent the work done in moving a particle from A to B along [3] three different paths 1, 2 and 3 respectively (as shown) in the gravitational field of a point mass m. Find the correct relation between W<sub>1</sub>, W<sub>2</sub> and W<sub>3</sub>



a)  $W_1 = W_2 = W_3$ 

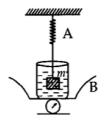
b)  $W_2 > W_1 > W_3$ 

c)  $W_1 > W_2 > W_3$ 

- d)  $W_1 < W_2 < W_3$
- 4. Which of the following sets have different dimensions?

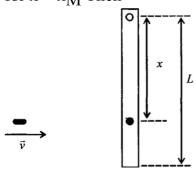
[3]

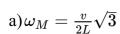
- a) Heat, Work done, Energy
- b) Pressure, Young's modulus, Stress
- c) Emf, Potential difference, Electric potential
- d) Dipole moment, Electric flux, Electric field
- 5. The spring balance A reads 2 kg with a block m suspended from it. A balance B [4] reads 5 kg when a beaker with liquid is put on the pan of the balance. The two balances are now so arranged that the hanging mass is inside the liquid in the beaker as shown in the figure. In this situation:



- a) the balance B will read more than 5 kg
- b) the balance A will read more than 2 kg
- c) the balance A and B will read 2 kg and 5 kg respectively
- d) the balance A will read less than 2 kg and B will read more than 5 kg

- 6. The coordinates of a particle moving in a plane are given by  $x(t) = a \cos(pt)$  and y [4] (t) = b  $\sin$  (pt) where a, b (< a) and p are positive constants of appropriate dimensions. Then
  - a) the acceleration of the particle is always directed towards a focus
- b) the path of the particle is an ellipse
- c) the velocity and acceleration of the particle are normal to each other at t $t=rac{\pi}{(2p)}$
- d) the distance travelled by the particle in time interval t = 0 to  $t=\frac{\pi}{(2p)}$  is a
- 7. A rod of mass m and length L, pivoted at one of its ends, is hanging vertically. A [4] bullet of the same mass moving at speed v strikes the rod horizontally at a distance x from its pivoted end and gets embedded in it. The combined system now rotates with angular speed  $\omega$  about the pivot. The maximum angular speed  $\omega_M$  is achieved for  $x = x_M$  Then





b)
$$x_M=rac{L}{\sqrt{3}}$$

$$\mathrm{c})\omega=rac{3vx}{L^2+3x^2}$$

$$\mathrm{d})\omega=rac{12vx}{L^2+12x^2}$$

8. The following field line can never represent



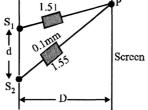
[4]

a) electrostatic field

- b) induced electric field
- c) magnetostatic field
- d) gravitational field of a mass at rest
- 9. A 20 cm long string having a mass of 1.0 gm is fixed at both the ends. The tension [4] in the string is 0.5 N. The string is set into vibration using an external vibrator of

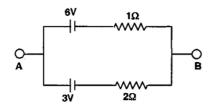
frequency 100 Hz. Find the separation (in cm) between the successive nodes on the string.

- 10. Gravitational acceleration on the surface of a planet is  $\frac{\sqrt{6}}{11}g$ , where g is the gravitational acceleration on the surface of the earth. The average mass density of the planet is  $\frac{2}{3}$  times that of the earth. If the escape speed on the surface of the earth is taken to be 11 km s<sup>-1</sup>, the escape speed on the surface of the planet in kms<sup>-1</sup> will be
- In a photoelectric experiment, a parallel beam of monochromatic light with power of [4] 200 W is incident on a perfectly absorbing cathode of work function 6.25 eV. The frequency of light is just above the threshold frequency so that the photoelectrons are emitted with negligible kinetic energy. Assume that the photoelectron emission efficiency is 100%. A potential difference of 500 V is applied between the cathode and the anode. All the emitted electrons are incident normally on the anode and are absorbed. The anode experiences a force  $F = n \times 10^{-4} N$  due to the impact of the electrons. The value of n is \_\_\_\_\_\_. Mass of the electron  $m_e = 9 \times 10^{-31} \text{ kg}$  and  $10\text{eV} = 1.6 \times 10^{-19} \text{ J}$ .
- 12. Consider an LC circuit, with inductance L = 0.1 H and capacitance  $C = 10^{-3}$ F, kept on a plane. The area of the circuit is  $1m^2$ . It is placed in a constant magnetic field of strength  $B_0$  which is perpendicular to the plane of the circuit. At time t = 0, the magnetic field strength starts increasing linearly as  $B = B_0 + \beta t$  with  $\beta = 0.04$  T s<sup>-1</sup>. The maximum magnitude of the current in the circuit is \_\_\_\_\_ mA.
- 13. In Young's double slit experiment, two slits  $S_1$  and  $S_2$  are **d** distance apart and the separation from slits to screen is D (as shown in figure). Now if two transparent slabs of equal thickness 0.1 mm but refractive index 1.51 and 1.55 are introduced in the path of beam ( $\lambda = 4000A$ ) from  $S_1$  and  $S_2$  respectively. The central bright fringe spot will shift by \_\_\_\_\_ number of fringes.

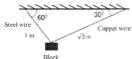


14. Two batteries of different emfs and different internal resistances are connected as shown. The voltage across AB in volts is





- 15. A 0.1 kg mass is suspended from a wire of negligible mass. The length of the wire is [4] 1m and its cross-sectional area is  $4.9 \times 10^{-7}$  m<sup>2</sup>. If the mass is pulled a little in the vertically downward direction and released, it performs simple harmonic motion of angular frequency 140 rad/sec. If the Young's modulus of the material of the wire is  $n \times 10^9$  Nm<sup>-2</sup>, then the value of n is:
- 16. A block of weight 100N is suspended by copper and steel wires of same crosssectional area  $0.5~{\rm cm}^2$  and, length  $\sqrt{3}{\rm m}$  and 1m, respectively. Their other ends are fixed on a ceiling as shown in figure. The angles subtended by copper and steel wires with the ceiling are  $30^{\circ}$  and  $60^{\circ}$ , respectively. If elongation in copper wire is and elongation in steel wire is  $(\Delta l_s)$ , then the ratio  $\frac{\Delta l_c}{\Delta l_s}$  is \_\_\_\_\_\_. [Young's modulus for copper and steel are  $1 \times 10^{11}~{\rm N/m}^2$  and  $2 \times 10^{11}~{\rm N/m}^2$ , respectively.]



# Chemistry

17. The correct set of quantum numbers for the unpaired electron of chlorine atom is [3]

a)	n	1	m	
	3	1	1	

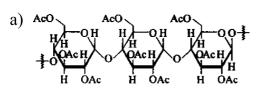
- 18. The rate of diffusion of methane at a given temperature is twice that of a gas X. The [3] molecular weight of X is
  - a)4.0

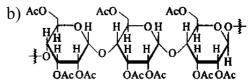
b)8.0

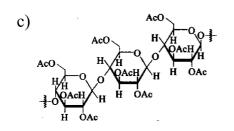
c)32.0

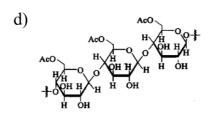
- d)64.0
- 19. Cellulose upon acetylation with excess acetic anhydride/ H<sub>2</sub>SO<sub>4</sub> (catalytic) gives cellulose triacetate whose structure is











20. Calamine, malachite, magnetite and cryolite, respectively, are

[3]

- $a) ZnCO_3, CuCO_3, Fe_2O_3, Na_3AlF_6b) ZnSO_4, Cu(OH)_2, Fe_3O_4, Na_3AlF_6b) ZnSO_5, Cu(OH)_5, Cu($
- $c)ZnSO_4, CuCO_3, Fe_2O_3, AlF_3$
- $d) ZnCO_3, CuCO_3 \cdot Cu(OH)_2, Fe_3O_4, Na_3AlF_6$
- 21. Only two isomeric monochloro derivatives are possible for

[4]

- a) 2-methylpropane
- b)benzene
- c)2,4-dimehylpentane
- d)n-butane

acid.

22. Which of the following statement(s) is (are) true?

[4]

- a) Monosaccharides cannot be hydrolysed to give poly
  - hydrolysed to give polyhydroxy aldehydes and ketones.

- c) Hydrolysis of sucrose gives dextrorotatory glucose and
- d) The two six-membered cyclic

b) Oxidation of glucose with

hemiacetal forms of D- (+)-

bromine water gives glutamic

- laevorotatory fructose. glucose are called anomers.
- 23. When NaNO<sub>3</sub> is heated in a closed vessel, oxygen is liberated and NaNO<sub>2</sub> is left behind. At equilibrium. [4]
  - a) addition of NaNO<sub>2</sub> favours reverse reaction
- b) addition of NaNO<sub>3</sub> favours forward reaction
- c) increasing pressure favours
- d) increasing temperature favours

reverse reaction

- forward reaction
- 24. The colour of the  $X_2$  molecules of group 17 elements changes gradually from yellow to violet down the group. This is due to

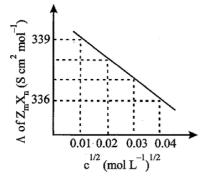


- a) Decrease in  $\pi^*$   $\sigma^*$  gap down the group
- b) The physical state of X<sub>2</sub> at room temperature changes from gas to solid down the group
- c) Decrease in ionization energy down the group
- d) Decrease in HOMO-LUMO gap down the group
- 25. Consider the strong electrolytes  $Z_m X_n$ ,  $U_m Y_p$  and  $V_m X_n$ . Limiting molar conductivity  $(\Lambda^0)$  of  $U_m Y_p$  and  $V_m X_n$  are 250 and 440 S cm<sup>2</sup> mol<sup>-1</sup>, respectively. The value of (m+n+p) is \_\_\_\_\_. Given:

Ion	$Z^{n+}$	U <sup>p+</sup>	V <sup>n+</sup>	x <sup>m</sup> -	Y <sup>m-</sup>
$\lambda^0$ (S cm <sup>2</sup> mol <sup>-1</sup> )	50.0	25.0	100.0	80.0	100.0

 $\lambda^0$  is the limiting molar conductivity of ions.

The plot of molar conductivity ( $\Lambda$ ) of  $Z_m X_n$  vs  $e^{\frac{1}{2}}$  is given below.



- 26. In the Arrhenius equation for a certain reaction, the value of A and  $E_a$  (activation energy) are  $4 \times 10^{13} \text{ sec}^{-1}$  and  $98.6 \text{ kJ mol}^{-1}$  respectively. If the reaction is of first order, at what temperature will its half-life period be ten minutes?
- 27. H<sub>2</sub>S (5 moles) reacts completely with acidified aqueous potassium permanganate solution. In this reaction, the number of moles of water produced is x, and the number of moles of electrons involved is y. The value of (x + y) is \_\_\_\_\_.
- 28. Consider the following molecules: Br<sub>3</sub>O<sub>8</sub>, F<sub>2</sub>O, H<sub>2</sub>S<sub>4</sub>O<sub>6</sub>, H<sub>2</sub>S<sub>5</sub>O<sub>6</sub>, and C<sub>3</sub>O<sub>2</sub>. [4] Count the number of atoms existing in their zero oxidation state in each molecule. Their sum is
- 29. To 500 cm<sup>3</sup> of water,  $3.0 \times 10^{-3}$  kg of acetic acid is added. If 23% of acetic acid is dissociated, what will be the depression in the freezing point  $K_f$  and density of water are 1.86 K kg<sup>-1</sup> mol<sup>-1</sup> and 0.997 g cm<sup>-3</sup>, respectively.

30. On dissolving 0.5 g of a non-volatile non-ionic solute to 39 g of benzene, its vapour [4] pressure decreases from 650 mm Hg to 640 mm Hg. The depression of freezing

point of Benzene (in K) upon addition of the solute is .

- (Given data: Molar mass and the molal freezing point depression constant of benzene are 78 g mol<sup>-1</sup> and 5.12 K kg mol<sup>-1</sup>, respectively)
- 31. The transformation occurring in Duma's method is given below: [4]  $C_2H_7N + \left(2x + \frac{y}{2}\right)CuO \rightarrow xCO_2 + \frac{y}{2}H_2O + \frac{z}{2}N_2 + \left(2x + \frac{y}{2}\right)Cu$  The value of y is \_\_\_\_\_\_. (Integer answer)
- 32. Among the following, the number of elements showing only one non-zero oxidation [4] state is:

O, Cl, F, N, P, Sn, Tl, Na, Ti

#### **Mathematics**

33. The number of real solutions of  $\tan^{-1} \sqrt{x(x+1)} + \sin^{-1} \sqrt{x^2 + x + 1} = \frac{\pi}{2} \text{ is}$ 

a)zero b)two

c)infinite d)one

34. For any positive integer n, define  $f_n:(0,\infty)\to\mathbb{R}$  as  $f_n(x)=$ 

 $\sum_{i=1}^{n} \tan^{-1} \left( \frac{1}{1 + (x+j)(x+j-1)} \right) \text{ for all } x \in (0, \infty)$ 

(Here, the inverse trigonometric function  $\tan^{-1} x$  assumes values in  $\left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$ )

Then, which of the following statement(s) is (are) TRUE?

- a) For any fixed positive integer n,  $\lim_{x \to \infty} \sec^2(f_n(x)) = 1$  b)  $\sum_{j=1}^{10} (1 + f_j(0)) \sec^2(f_j(0)) = 10$
- c) For any fixed positive integer n,  $\lim_{x \to \infty} \tan(f_n(x)) = \frac{1}{n}$  d)  $\sum_{j=1}^{5} \tan^2(f_j(0)) = 55$
- 35. If the system of equations x + ay = 0, az + y = 0 and ax + z = 0 has infinite solutions, then the value of a is

a) 1 b) -1

c)0 d)no real values

36. The function  $f(x) = \frac{\log(1+ax) - \log(1-bx)}{x}$  is not defined at x = 0. The value which should be assigned to f at x = 0, so that it is continuous at x = 0, is

$$a)a + b$$

b) log a - log b

 $d)\log a + \log b$ 

37. Three lines  $L_1: \vec{r} = \lambda \hat{i}, \lambda \in R, L_2: \vec{r} = \hat{k} + \mu \hat{j}, \mu \in R$  and  $L_3: \vec{r} = \hat{i} + \hat{j} + v\hat{k}, v \in R$  are given. For which point(s) Q on  $L_2$  can we find a point P on  $L_1$  and a point R on  $L_3$  so that P, Q and R are collinear?

a) 
$$\hat{k} + \hat{j}$$

 $b)\hat{k}$ 

$$c)\hat{k} + \frac{1}{2}\hat{j}$$

 $\mathrm{d})\hat{k} - rac{1}{2}\hat{j}$ 

38. Let a hyperbola passes through the focus of the ellipse  $\frac{x^2}{25} + \frac{y^2}{16} = 1$ . The transverse and conjugate axes of this hyperbola coincide with the major and minor axes of the given ellipse, also the product of eccentricities of given ellipse and hyperbola is 1, then

a) the equation of hyperbola is

b) vertex of hyperbola is  $(5\sqrt{3}, 0)$ 

$$\frac{x^2}{9} + \frac{y^2}{16} = 1$$

c) focus of hyperbola is (5, 0)

d) the equation of hyperbola is

$$\frac{x^2}{9} + \frac{y^2}{25} = 1$$

39. A solution curve of the differential equation  $(x^2 + xy + 4x + 2y + 4) \frac{dy}{dx} - y^2 = 0$ , x > [4]

0, passes through the point (1, 3). Then the solution curve

a) does NOT intersect  $y = (x + 3)^2$ 

b) intersects y = x + 2 exactly at

two points

c) intersects y = x + 2 exactly at d) intersects  $y = (x + 2)^2$  one point

40. There are three bags  $B_1$ ,  $B_2$ , and  $B_3$ . The bag  $B_1$  contains 5 red and 5 green balls,  $B_2$  contains 3 red and 5 green balls, and  $B_3$  contains 5 red and 3 green balls. Bags  $B_1$ ,  $B_2$  and  $B_3$  have probabilities  $\frac{3}{10}$ ,  $\frac{3}{10}$  and  $\frac{4}{10}$  respectively of being chosen. A bag is selected at random and a ball is chosen at random from the bag. Then which of the following options is/are correct?

- a) Probability that the chosen ball is green, given that the selected bag is  $B_3$ , equals  $\frac{3}{8}$
- b) Probability that the selected bag is  $B_3$  and the chosen ball is green equals  $\frac{3}{10}$
- c) Probability that the selected bag is B<sub>3</sub>, given that the chosen ball is green, equals  $\frac{5}{13}$
- d) Probability that the chosen ball is green equals  $\frac{39}{80}$
- 41. Let M be a 3 × 3 matrix satisfying  $M \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix} = \begin{bmatrix} -1 \\ 2 \\ 3 \end{bmatrix}$ ,  $M \begin{bmatrix} 1 \\ -1 \\ 0 \end{bmatrix} = \begin{bmatrix} 1 \\ 1 \\ -1 \end{bmatrix}$ , and  $M \begin{bmatrix} 1 \\ 1 \\ 1 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 12 \end{bmatrix}$

Then, the sum of the diagonal entries of M is

- 42. For  $x \in \mathbb{R}$ , then number of real roots of the equation  $3x^2 4|x^2 1| + x 1 = 0$  is [4]
- 43. The value of  $\left((\log_2 9)^2\right)^{\frac{1}{\log_2(\log_2 9)}} \times (\sqrt{7})^{\frac{1}{\log_4 7}}$  is \_\_\_\_\_.
- 44. Let f be a function defined on R (the set of all real numbers) such that f'(x) = 2010 [4]  $(x 2009)(x 2010)^2(x 2011)^3(x 2012)^4$ ,  $\forall x \in R$ . If g is a function defined on R with values in the interval  $(0, \infty)$  such that f(x) = In (g(x)),  $\forall x \in R$ , then the number of points in R at which g has a local maximum is
- 45. Let a and b be two non-zero real numbers. If the coefficient of  $x^5$  in the expansion of  $\left(ax^2 + \frac{70}{27bx}\right)^4$  is equal to the coefficient of  $x^{-5}$  in the expansion of  $\left(ax \frac{1}{bx^2}\right)^7$ , then the value of 2b is
- 46. Consider the lines  $L_1$  and  $L_2$  defined by  $L_1 \colon x\sqrt{2} + y 1 = 0 \text{ and } L_2 \colon x\sqrt{2} y + 1 = 0$  For a fixed constant  $\lambda$ , let C be the locus of a point P such that the product of the distance of P from  $L_1$  and the distance of P from  $L_2$  is  $\lambda_2$ . The line y = 2x + 1 meets C at two points R and S, where the distance between R and S is  $\sqrt{270}$ . Let the perpendicular bisector of RS meet C at two distinct points R' and S'. Let D be the **square** of the distance between R' and S'. The value of  $\lambda^2$  is \_\_\_\_\_\_.
- 47. Let  $f: R \to R$  be a differentiable function such that its derivative f' is continuous [4] and  $f(\pi) = -6$ .

If F:  $[0, \pi] \to R$  is defined by  $F(x) = \int_0^x f(t)dt$ , and if  $\int_0^x (f'(x) + F(x)) \cos x \, dx = 2$ , then the value of f(0) is \_\_\_\_\_.

48. Let  $a_1$ ,  $a_2$ ,  $a_3$ ,...,  $a_{100}$  be an arithmetic progression with  $a_1 = 3$  and  $S_p$  [4]  $= \sum_{i=1}^p a_i, 1 \le p \le 100.$  For any integer n with  $1 \le n \le 20$ , let m = 5n. If  $\frac{S_m}{S_n}$  does not depend on n, then  $a_2$  is equal to

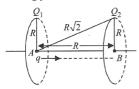
#### **Solution**

# **Physics**

1. (a) 
$$\frac{q(Q_1-Q_2)(\sqrt{2}-1)}{(4\sqrt{2}\pi\varepsilon_0 R)}$$

#### **Explanation:**

The work done in moving a charge q from centre of one ring to other = change in potential energy



$$\begin{split} & \mathbf{W}_{\mathbf{A}\mathbf{B}} = \mu \\ & \mu_A = \left[ \left( \frac{Q_1}{4\pi\varepsilon_0 R} \right) \times q + \left( \frac{Q_2}{4\pi\varepsilon_0 \sqrt{R^2 + R^2}} \right) q \right] \\ & = \frac{q}{4\pi\varepsilon_0 R} \left[ Q_1 + \frac{Q_2}{\sqrt{2}} \right] \\ & \mu_B = \left[ \left( \frac{Q_2}{4\pi\varepsilon_0 R} \right) q + \left( \frac{Q_1}{4\pi\varepsilon_0 \sqrt{R^2 + R^2}} \right) q \right] \\ & = \frac{q}{4\pi\varepsilon_0 R} \left[ Q_2 + \frac{Q_1}{\sqrt{2}} \right] \\ & \therefore W_{AB} = \frac{q}{4\pi\varepsilon_0 R} \left[ Q_1 + \frac{Q_2}{\sqrt{2}} - Q_2 - \frac{Q_1}{\sqrt{2}} \right] \\ & = \frac{q(Q_1 - Q_2)}{4\pi\varepsilon_0 R} \left( \frac{\sqrt{2} - 1}{\sqrt{2}} \right) \end{split}$$

# 2. **(a)** $\frac{x}{8}$

#### **Explanation:**

The magnetic field at the centre of a circular loop of radius R carrying current I is,

$$B=rac{\mu_0 2\pi I}{4\pi R}=rac{\mu_0 I}{2R}$$

Its magnetic moment is,  $M = IA = I(\pi R^2)$ 

$$\therefore \frac{B}{M} = \frac{\mu_0 I}{2R} \times \frac{1}{l\pi R^2} = \frac{\mu_0}{2\pi R^3} = x \text{ (Given)}$$

When both the current and radius is doubled, the ratio becomes

$$\therefore \frac{B'}{M'} = \frac{\mu_0}{2\pi (2R)^3} = \frac{1}{8} \left( \frac{\mu_0}{2\pi R^3} \right) = \frac{x}{8}$$

3. (a) 
$$W_1 = W_2 = W_3$$

# **Explanation:**

Gravitational field is a conservative force field. In a conservative force field work done is path independent

$$\therefore \mathbf{W}_1 = \mathbf{W}_2 = \mathbf{W}_3$$

4.

(d) Dipole moment, Electric flux, Electric field





# **Explanation:**

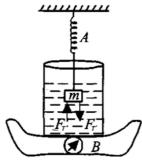
Dipole moment =  $(charge) \times (distance)$ 

Electric flux = (electric field)  $\times$  (area)

Hence, the correct option is (d).

- 5. (a) the balance B will read more than 5 kg
  - (d) the balance A will read less than 2 kg and B will read more than 5 kg

**Explanation:** When the block of mass m is inside the liquid an upthrust  $F_T$  will act on the mass which will decrease the reading on A i.e., will read less than 2 kg.



According to Newton's third law, to each and every action, there is equal and opposite reaction.

So by reaction an equal force will be exerted on the liquid of the beaker which will increase the reading in B i.e., will read more than 5 kg.

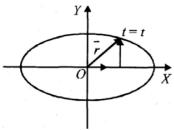
- 6. (a) the acceleration of the particle is always directed towards a focus
  - (b) the path of the particle is an ellipse
  - (c) the velocity and acceleration of the particle are normal to each other at  $tt = \frac{\pi}{(2p)}$

**Explanation:**  $x = a \cos pt \Rightarrow \cos (pt) = \frac{x}{a} ...(i)$ 

$$y = b \sin pt \Rightarrow \sin (pt) = \frac{y}{b}$$
 ...(ii)

Squaring and adding eqn. (i) and (ii), we get,

$$\frac{x^2}{a^2} + \frac{y^2}{b^2} = 1$$



Hence path of the particle is an ellipse.

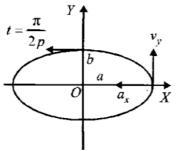
From the given equations we can find,

$$\frac{dx}{dt} = v_X = -ap \sin pt; \frac{d^2x}{dt^2} = a_X = -ap^2 \cos pt$$

$$\frac{dy}{dt} = v_y = \text{pb cos pt}$$

and 
$$\frac{d^2y}{dt^2} = a_y = -bp^2 \sin pt$$





At time  $t = \frac{\pi}{2p}$  or  $pt = \frac{\pi}{2}$ 

 $a_X$  and  $v_V$  become zero (because  $\cos\frac{\pi}{2}=0$ ). Only  $v_X$  and  $a_Y$  are left, or we can say that velocity is along negative x-axis and acceleration along negative y-axis.

Hence, at  $t = \frac{\pi}{2n}$ , velocity and acceleration of the particle are normal to each other.

At t = t, position of the particle

$$ec{r}(t) = x \hat{i} + y \hat{j} = a \cos p t \hat{i} + b \sin p t \hat{j}$$

and acceleration of the particle

$$egin{aligned} ec{a}(t) &= a_x \hat{i} \, + a_y \hat{j} = -p^2 [a\cos pt \hat{i} \, + b\sin pt \hat{j}] \ &= -p^2 [x \hat{i} \, + y \hat{j}] = -p^2 ec{r}(t) \end{aligned}$$

Therefore, acceleration of the particle is always directed towards origin.

At t = 0, particle is at (a, 0) and at  $t = \frac{\pi}{2p}$ , particle is at (0, b). Therefore, the distance covered is one fourth of the elliptical path and not a.

7. (a) 
$$\omega_M = \frac{v}{2L}\sqrt{3}$$

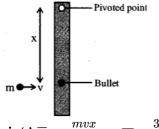
(b) 
$$x_M = rac{L}{\sqrt{3}}$$

(b) 
$$x_M = \frac{L}{\sqrt{3}}$$
  
(c)  $\omega = \frac{3vx}{L^2 + 3x^2}$ 

Explanation: From angular momentum conservation about the pivoted point.

$$\mathrm{mvx} = \left(rac{\mathrm{mL}^2}{3} + \mathrm{mx}^2
ight) \omega$$

[As the combined system rotates with angular speed  $\omega$  about the pivot]



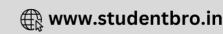
$$\therefore \omega = rac{mvx}{rac{mL^2}{mL^2} + mx^2} = rac{3vx}{L^2 + 3x^2}$$
  $\omega = rac{3vx}{L^2 + 3x^2}$ 

Hence option (
$$\omega = \frac{3vx}{L^2 + 3x^2}$$
) is correct.

For maximum angular velocity,  $\frac{d\omega}{dx} = 0$ 

$$egin{array}{l} rac{\mathrm{d}}{\mathrm{dx}} \left(rac{\mathrm{L}^2}{\mathrm{x}} + 3\mathrm{x}
ight) = 0 \ \Rightarrow rac{\mathrm{L}^2}{x^2} + 3 = 0 \Rightarrow x = rac{\mathrm{L}}{\sqrt{3}} \end{array}$$





$$\therefore X_{\mathrm{m}} = \frac{L}{\sqrt{3}}$$

So option  $(x_M = \frac{L}{\sqrt{3}})$  is correct.

$$\omega_m=rac{3vx}{L^2+3x^2}=rac{3vrac{L}{\sqrt{3}}}{L^2+3\left(rac{L}{\sqrt{3}}
ight)^2}=rac{\sqrt{3}}{2L}V$$

Hence option  $(\omega_M = \frac{v}{2L}\sqrt{3})$  is correct.

- 8. (a) electrostatic field
  - (d) gravitational field of a mass at rest

**Explanation:** Only induced electric field and magnetostatic field form closed loops of field lines.

9.5

Explanation:

Wave speed = 
$$\sqrt{\frac{T}{\mu}} = \sqrt{\frac{0.5}{10^{-3}/0.2}} = 10 \text{ m/s}$$

Using, 
$$v = f\lambda$$

$$\lambda = \frac{v}{f} = \frac{10}{100} = 10 \text{ cm}$$

Distance between successive nodes =  $\frac{\lambda}{2}$  = 5 cm

10.3

Explanation:

$$g=rac{GM}{R^2}=rac{G\left(rac{4}{3}\pi R^3
ight)
ho}{R^2} ext{ or } g \propto 
ho R ext{ or } R \propto rac{g}{
ho}$$

Now escape velocity,  $v_e=\sqrt{2gR}$ 

or 
$$v_e \propto \sqrt{gR}$$
 or,  $v_e \propto \sqrt{g imes rac{g}{
ho}} \propto \sqrt{rac{g^2}{
ho}}$ 

:. 
$$(v_e)_{planet} = (11 \text{ kms}^{-1}) \sqrt{\frac{6}{121} \times \frac{3}{2}} = 3 \text{ kms}^{-1}$$

- ... The correct answer is 3
- 11.24

**Explanation:** 

Number of electrons emitted/s = 
$$\frac{200W}{6.25 \times 1.6 \times 10^{-19} J}$$

Force, F = Rate of change of linear momentum =  $N\sqrt{2mV}$ 

$$= \frac{200}{6.25 \times 1.6 \times 10^{-19}} \times \sqrt{2 \times 9 \times 10^{-31} \times 1.6 \times 10^{-19} \times 500}$$
  

$$\Rightarrow F = 24 \times 10^{-4} \text{ N } [\because K = \text{eV: e} = 1.6 \times 10^{-19} = \text{V} = 500]$$

$$\Rightarrow$$
 F = 24 × 10<sup>-4</sup> N [: K = eV: e = 1.6 × 10<sup>-19</sup> = V = 500]

- $\therefore$  n = 24
- 12.4.0

**Explanation:** 

Induced Emf,

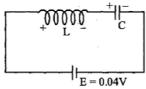
$$\mathrm{E} = \left| rac{d\phi}{dt} 
ight| = rac{d}{dt} (BA) = rac{Ad}{dt} [(B_0 + eta t)]$$

$$= A\beta = 1 \times 0.04 = 0.04 \text{ volt}$$





So, circuit can be drawn as



$$E = 0.04V$$
By KVL,  $E = L \frac{di}{dt} + \frac{q}{c}$ 

$$\Rightarrow L \frac{di}{dt} = E - \frac{q}{c}$$

$$\Rightarrow \frac{d^2q}{dt^2} = -\frac{1}{LC}(q - CE)$$

Comparing it with equation of SHM, we get

$$q = CE + A \sin(\omega t + \phi)$$
, where  $\omega = \frac{1}{\sqrt{LC}}$ 

So, 
$$i = A \omega \cos(\omega t + \phi)$$

at 
$$t = 0$$
,  $q = 0$  and  $i = 0$ 

So, 
$$O = CE + A \sin \phi \Rightarrow A \sin \phi = -CE ...(i)$$

$$O = \cos \phi \Rightarrow \phi = \frac{\pi}{2}$$
 ...(ii)

from (i) and (ii), we get

$$A = -CE$$

So, 
$$i = -CE \omega \cos(\omega t + \frac{\pi}{2})$$

= CE 
$$\omega \sin \omega t$$

Therefore, 
$$i_{\text{max}} = \text{CE } \omega = 10^{-3} \times 0.04 \times \frac{1}{\sqrt{0.1 \times 10^{-3}}} = 4 \text{ mA}$$

## 13. 10.0

**Explanation:** 

Path difference at P be  $\Delta x$ 

$$\Delta x = (\mu_2 - \mu_1)t = (1.55 - 1.51)0.1 \text{ mm}$$

$$= 0.04 \times 10^{-4}$$

$$\Delta x = 4 \times 10^{-6} = 4 \ \mu \text{m}$$

Here, y is the distance of central maxima from geometric center

$$y = \frac{\Delta xD}{d} = 4 \times 10^{-6} \frac{D}{d}$$

fringe width 
$$(\beta) = \frac{\lambda D}{d} = 4 \times 10^{-6} \frac{D}{d} = 4 \frac{D}{d} \mu m$$

... Central bright fringe spot will shift by 'x'

Number of shift 
$$=\frac{y}{\beta}=rac{4 imes10^{-6}rac{D}{d}}{4 imes10^{-7}rac{D}{d}}=10$$

#### 14. 5.0

**Explanation:** 

Let i be the current flowing in the circuit. Applying Kirchhoff's law KVL in the loop

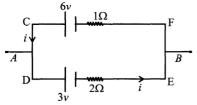
**CDEFC** 

$$-3 - 2i - i + 6 = 0 \Rightarrow 3i = 3$$

$$\therefore$$
 i = 1 A



Now for upper or lower path



$$V_{A} - 6 + 1 \times 1 = V_{B}$$

$$\therefore$$
 V<sub>A</sub> - V<sub>B</sub> = 5V

or, 
$$V_A - 3 - 2 \times 1 = V_B$$

$$\therefore$$
 V<sub>A</sub> - V<sub>B</sub> = 5 V

15.4

Explanation:

Let Y is the Young's modulus of wire. In equilibrium, the tension in the wire is,

$$T = mg = 0.1 \times 9.8 = 0.98 \text{ N}$$

Let elongation in wire in equilibrium is I<sub>0</sub>, then,

$$m Y = rac{T/A}{l_0L} = rac{0.98 imes 1}{4.9 imes 10^{-7} imes L_0} = rac{2 imes 10^6}{l_0}$$

Let us say the wire is pulled by distance x from its equilibrium position, then in this displaced position, let F is the tension in the wire, then,

$$ext{Y} = rac{F/A}{(l_0+x)/L} = rac{FL}{A(l_0+x)}$$
 or  $ext{F} = rac{Y(l_0+x)A}{L}$ 

Restoring force on block the s,  $F_{res} = F - mg$ 

$$F_{res} = \frac{Yl_0A}{L} + \frac{Y_{xA}}{L} - mg$$

$$= \frac{YA}{L} \times x \left[\because \frac{Yl_0A}{L} = mg\right]$$

$$\therefore a = \frac{YA}{mL}x$$

As a  $\propto$  x, so block performs SHM with time period given by:

$$T = \frac{2\pi}{\sqrt{\frac{YA}{mL}}} \text{ and } w = \frac{2\pi}{T} = \sqrt{\frac{YA}{mL}}$$

$$\therefore (140)^{2} = \frac{YA}{mL} = \frac{Y \times 4.9 \times 10^{-7}}{0.1 \times 1}$$

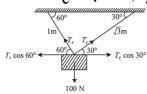
$$\therefore Y = 4 \times 10^{9} \text{ N/m}^{2}$$

Hence, the required answer is, n = 4

16. 2

Explanation:

Given: 
$$l_c = \sqrt{3}$$
 m;  $l_s = 1$  m;  $Y_c = 1 \times 10^{11}$  N/m<sup>2</sup> and  $T = 2 \times 10^{11}$  N/m<sup>2</sup>.



At equilibrium,  $T_S \cos 60^\circ = T_C \cos 30^\circ$ 

$$\Rightarrow rac{T_S}{2} = rac{T_c\sqrt{3}}{2} \Rightarrow T_3 = \sqrt{3}T_r \Rightarrow rac{T_c}{T_s} = rac{1}{\sqrt{3}}$$

$$\therefore \frac{l_c}{l_s} = \frac{\sqrt{3}}{1} \text{ and } \frac{Y_c}{Y_s} = \frac{1 \times 10^{11}}{2 \times 10^{11}} = \frac{1}{2}$$
From,  $Y = \frac{Fl}{A\Delta l} \Rightarrow \Delta l = \frac{Fl}{AY}$ 

From, Y = 
$$\frac{Fl}{A\Delta l}$$
  $\Rightarrow$   $\Delta l = \frac{Fl}{AY}$ 

Here,  $A_S = A_C$ 

$$\therefore \frac{\Delta l_c}{\Delta l_s} = \left(\frac{T_c}{T_s}\right) \times \left(\frac{l_c}{l_s}\right) \times \left(\frac{Y_s}{Y_c}\right) = \left(\frac{1}{\sqrt{3}}\right) \times \left(\frac{\sqrt{3}}{1}\right) \times \left(\frac{2}{1}\right) = 2$$

#### 17. **(a)**

n	1	m
3	1	1

## **Explanation:**

$$C1(17) = Is^2 2s^2 2p^6 3s^2 3p^5$$



The last, unpaired electron has, n = 3, l = 1(p) and m can have any of the three value (-1,

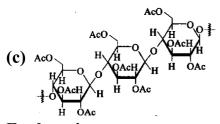
18.

**(d)** 64.0

## **Explanation:**

$$rac{r_{ ext{CH}_4}}{r_X}=2=\sqrt{rac{M_X}{16}}\Rightarrow M_X$$
 = 64

19.



## **Explanation:**

20.

(d)  $ZnCO_3$ ,  $CuCO_3 \cdot Cu(OH)_2$ ,  $Fe_3O_4$ ,  $Na_3AlF_6$ 

#### **Explanation:**

 $ZnCO_3$ ,  $CuCO_3 \cdot Cu(OH)_2$ ,  $Fe_3O_4$ ,  $Na_3AlF_6$ 



- 21. (a) 2-methylpropane
  - (d) n-butane

#### **Explanation:**

- a. In n-butane, Cl can add at either the first or second carbon giving two isomers.
- b.  $\rm CH_3-C_{H_3}-C_{H_3}-C_{H_3}$  three isomers with Cl group at either of the CH  $_3$   $_{\rm CH_3}$

groups, second C-atom and third C-atom.

- c. Benzene forms only one single derivative.
- d.  ${\rm CH_3-C_{H_3}}$  Will again give two isomers with Cl at either one of the CH3  $_{\rm CH_3}$

groups or at the central C-atom.

- 22. (a) Monosaccharides cannot be hydrolysed to give poly-hydroxy aldehydes and ketones.
  - (c) Hydrolysis of sucrose gives dextrorotatory glucose and laevorotatory fructose.
  - (d) The two six-membered cyclic hemiacetal forms of D- (+)- glucose are called anomers.

#### **Explanation:**

a. Glucose  $\xrightarrow{\text{Br}_2}$  Gluconic acid.

Bromine water oxidises only aldehyde group to carboxylic group. It neither oxidises - OH group nor -CO group.

 $b. \ \, \underbrace{\text{Sucrose} \xrightarrow{\text{H}_2\text{O}}}_{(+)} \text{Glucose} \ + \ \, \underbrace{\text{Fructose}}_{(-)}$ 

Sucrose is dextrorotatory  $(+66.6^{\circ})$  in nature but on hydrolysis it gives dextrorotatory glucose  $(+52.5^{\circ})$  and laevorotatory fructose  $(-92.4^{\circ})$ . Overall solution is laevorotatory since laevorotation is more than dextrorotation.

c.

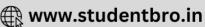
Because they differ in configuration only around C<sub>1</sub> position

- d. Monosaccharides do not undergo further hydrolysis due to absence of glycosidic linkages.
- 23. (c) increasing pressure favours reverse reaction
  - (d) increasing temperature favours forward reaction

**Explanation:**  $2NaNO_3(s) \rightleftharpoons 2NaNO_2(s) + O_2(g)$ 

According to Le-Chatelier principle, an increase in pressure always favours the reaction, where volume or moles decrease (i.e. reverse direction). As heat is added, i.e. reaction is





endothermic and is supported in forward direction with increase in temperature. NaNO3 and NaNO<sub>2</sub> both are solid. Thus, they will not effect the position of equilibrium.

- 24. (a) Decrease in  $\pi^*$   $\sigma^*$  gap down the group
  - (d) Decrease in HOMO-LUMO gap down the group

**Explanation:** Energy,  $E = \frac{hc}{\lambda}$ 

On moving down the group, the colour of the X<sub>2</sub> molecule of group 17 elements changes gradually from yellow to violet. This happens because the amount of energy required for the excitation of the halogen atom decreases down the group. HOMO ( $\pi^*$ ) - LUMO (  $\sigma^*$ ) gap decreases down the group that makes  $\pi^*$  to  $\sigma^*$  excitation easier. Lesser the energy gap, more is the wavelength of light absorbed and hence, lesser is the wavelength of light emitted.

#### 25.7

**Explanation:** 

Explanation: 
$$\Lambda^{\circ}(U_{m}Y_{p}) = m\lambda^{\circ}(U^{P+}) + p\lambda^{\circ}(Y^{m-})$$
  $\Rightarrow 25m + 100p = 250$   $\Rightarrow m + 4p = 10 \dots (1)$  
$$\Lambda^{\circ}(V_{m}Xn) = m\lambda^{\circ}(V^{n+}) + n\lambda^{\circ}(X^{m-})$$
  $\Rightarrow 100m + 80n = 440$   $\Rightarrow 5m + 4n = 22 \dots (2)$  For electrolyte  $Z_{m}X_{n}$  from the given curve, 
$$\Lambda(Z_{m}X_{n}) = \Lambda^{\circ}(Z_{m}X_{n}) - A\sqrt{C}$$
 Slope,  $m = -A = \frac{339 - 336}{0.01 - 0.04} \Rightarrow A = 100$  For  $\lambda_{m} = 339 \text{ S cm}^{2} \text{ mol}^{-1}$ ,  $\sqrt{C} = 0.01 (\text{mol } L - 1)^{\frac{1}{2}}$   $339 = \Lambda^{\circ}(Z_{m}X_{n}) - 100 \times 0.01$   $\Rightarrow \Lambda^{\circ}(Z_{m}X_{n}) = 340 \text{ S cm}^{2} \text{ mol}^{-1}$   $\Rightarrow m\lambda^{\circ}(Z^{n+}) + n\lambda^{\circ}(Z^{m-}) = 340$   $\Rightarrow 50m + 80n = 340$   $\Rightarrow 5m + 8n = 34 \dots (3)$  From eq. (2) and (3),  $n = 3$  and  $m = 2$  Putting value of m in eq. (1),

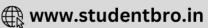
Therefore, m + n + p = 2 + 3 + 2 = 7

26.311.35

p = 2

Explanation:





According to Arrhenius equation

$$\log k = \log A - \frac{E_a}{2.303 RT}$$

log k = log A - 
$$\frac{E_a}{2.303RT}$$
  
We know that k =  $\frac{0.693}{t_{\frac{1}{2}}}$  =  $\frac{0.693}{10\times60}$  ( $t_{\frac{1}{2}}$  = 10 × 60 sec)

$$= 1.555 \times 10^{-3}$$

Substituting the various values in the above equation, we get

$$\log 1.155 \times 10^{-3} = \log 4 \times 10^{13} - \frac{98.6}{2.303 \times 8.314 \times 10^{-3} \times T}$$

On usual calculations, T = 311.35 K

#### 27.18

**Explanation:** 

$$2~KMnO_{4}^{+7} + 5H_{2}~S + 3H_{2}SO_{4} \rightarrow K_{2}SO_{4} + \overset{+2}{2}~MnSO_{4} + 5~S + 8H_{2}O$$

x = 8 (moles of H<sub>2</sub>O produced)

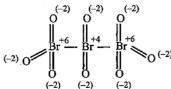
$$y = 14 - 4 = 10$$
 (number of electrons involved)

$$x + y = 10 + 8 = 18$$

#### 28.6

**Explanation:** 

$$Br_3, O_8$$



Number of atoms with zero oxidation state = 0

Number of atoms with zero oxidation state = 0

# H2S<sub>4</sub>O<sub>6</sub>

$$(+1) \quad (-2) \quad | S + 5 \atop H - O \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O - H \quad | S - 5 \atop O -$$

Number of atoms with zero oxidation state = 2

$$H_2S_5O_6$$

$$H-O = S^{+5} S^{0} - S^{0} - S^{0} = S^{0} - H$$

Number of atoms where zero oxidation state = 3

 $C_3O_2$ 



Number of atoms with zero oxidation state = 1

$$Sum = 2 + 3 + 1 = 6$$

29. 0.228

Explanation:

$$\Delta T_f = i imes K_f imes m$$

Weight of water =  $500 \times 0.997 = 498.85$  g (weight = volume × density)

No. of moles of acetic acid

$$=\frac{\mathrm{Wt.~of~CH_3COOH~in~g}}{\mathrm{Mol.~wt.~of~CH_3COOH}}=\frac{3\times10^{-3}\times10^{3}}{60}=0.05$$

Since 498.5 g of water has 0.05 moles of CH<sub>3</sub>COOH

1000 g of water has = 
$$\frac{0.05 \times 1000}{498.5} = 0.1$$

Therefore, molality of the solution = 0.1

Determination of van't Hoff factor, i

$$\begin{array}{c} CH_3COOH \longrightarrow CH_3COO^+ + \ H^+ \\ No. \ of \ moles \ at \ start & 1 & 0 & 0.23 \\ No. \ of \ moles \ at \ equb. & 1-0.23 & 0.23 & 0.23 \end{array}$$

Therefore, van't Hoff factor

$$= \frac{\textit{No. of particles after dissociation}}{}$$

 $No.\ of\ particles\ before\ dissociation$ 

$$=\frac{1-0.23+0.23+0.23}{1}=1.23$$

Now, we know that,

$$\Delta T_f = i \times K_f \times m = 1.23 \times 1.86 \times 0.1 = 0.228 \text{ K}$$

30. 1.03

**Explanation:** 

We know that,

$$\begin{split} \frac{P^0 - P}{P^0} &= \frac{w_2 \times m_1}{m_2 \times w_1} \\ \frac{10}{640} &= \frac{0.5 \times 78}{m_2 \times 39} \\ m_2 &= 64g \\ \Delta T_f &= \frac{K_f \times w_2 \times 1000}{m_2 \times w_l} \\ \Delta T_f &= \frac{5.12 \times 0.5 \times 1000}{64 \times 39} = 1.0256 \approx 1.03 \end{split}$$

31. 7.0

**Explanation:** 

$$C_2H_7N + \left(2x + \frac{y}{2}\right)CuO \longrightarrow xCO_2 + \frac{y}{2}H_2O + \frac{z}{2}N_2 + \left(2x + \frac{y}{2}\right)Cu ...(i)$$

$$C_xH_yN_z + \left(2x + \frac{y}{2}\right)CuO \longrightarrow xCO_2 + \frac{y}{2}H_2O + \frac{z}{2}N_2 + \left(2x + \frac{y}{2}\right)Cu ...(ii)$$

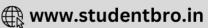
Comparing (i) & (ii), we get: x = 2, y = 7, z = 1

32. 2

Explanation:

Fluorine generally shows 0 and -1 oxidation states while sodium shows 0 and +1 oxidation state.





33.

**(b)** two

# **Explanation:**

Given function is

$$an^{-1}\sqrt{x(x+1)}+\sin^{-1}\sqrt{x^2+x+1}=rac{\pi}{2}$$

Function is defined, if

- i.  $x(x+1) \ge 0$ , since domain of square root function.
- ii.  $x^2 + x + 1 \ge 0$ , since domain of square root function.
- iii.  $\sqrt{x^2 + x + 1} \le 1$ , since domain of sin<sup>-1</sup> function.

From (ii) and (iii),  $0 \le x^2 + x + 1 \le 1 \cap x^2 + x \ge 0$ 

$$\Rightarrow 0 \leq x^2 + x + 1 \leq 1 \cap x^2 + x + 1 \geq 1$$

$$\Rightarrow x^2 + x + 1 = 1$$

$$\Rightarrow x^2 + x = 0$$

$$\Rightarrow$$
 x(x + 1) = 0

$$\Rightarrow$$
 x = 0, x = -1

34. (a) For any fixed positive integer n,  $\lim_{n \to \infty} \sec^2(f_n(x)) = 1$ 

# **Explanation:**

$$f_{\mathbf{n}}(\mathbf{x}) = \sum_{j=1}^{n} \tan^{-1} \left( \frac{1}{1 + (x+j)(x+j-1)} \right)$$

$$= \sum_{j=1}^{n} \tan^{-1} \left[ \frac{(x+j)-(x+j-1)}{1+(x+j)(x+j-1)} \right]$$

$$= \sum_{i=1}^{n} [\tan^{-1} (x+j) - \tan^{-1} (x+j-1)]$$

$$\Rightarrow f_n(x) = \tan^{-1}(x+n) - \tan^{-1}(x)$$

$$= \tan^{-1} \left( \frac{n}{1 + x(n+x)} \right)$$

$$\Rightarrow f_{1}(x) = \frac{1}{1 + (x+n)^2} - \frac{1}{1 + x^2}$$

$$\Rightarrow f_n(x) = \frac{1}{1 + (x+n)^2} - \frac{1}{1 + x^2}$$
and  $f_n(0) = \tan^{-1}(n)$ ,  $\therefore \tan^2(\tan^{-1}n) = n^2$ 

Here x = 0 is not in the given domain, i.e.,  $x \in (0, \infty)$ 

$$\lim_{x \to \infty} \sec^2(f_n(x)) = \lim_{x \to \infty} 1 + \tan^2(f_n(x))$$

$$= 1 + \lim_{x \to \infty} \tan^2(f_n(x)) = 1$$

35.

**(b)** -1

# **Explanation:**

Given equations x + ay = 0, az + y = 0, ax + z = 0 has infinite solutions.

$$\begin{vmatrix} 1 & a & 0 \\ 0 & 1 & a \\ a & 0 & 1 \end{vmatrix} = 0$$

$$\Rightarrow 1 + a^3 = 0 \text{ or } a = -1$$

36. (a) 
$$a + b$$

# **Explanation:**

For f(x) to be continuous, we must have

$$f(0) = \lim_{x \to 0} f(x)$$

$$= \lim_{x \to 0} \frac{\log(1 + ax) - \log(1 - bx)}{x}$$

$$= \lim_{x \to 0} \frac{a \log(1 + ax)}{ax} + \frac{b \log(1 - bx)}{-bx}$$

$$= a \cdot 1 + b \cdot 1 \quad \left[ \text{ using } \lim_{x \to 0} \frac{\log(1 + x)}{x} = 1 \right]$$

$$= a + b$$

$$\therefore f(0) = (a + b)$$

37. **(c)** 
$$\hat{k} + \frac{1}{2}\hat{j}$$
 **(d)**  $\hat{k} - \frac{1}{2}\hat{j}$ 

**Explanation:** Let any point  $P(\lambda, 0, 0)$  on  $L_1$ ,  $Q(0, \mu, 1)$  on  $L_2$  and R(1, 1, v) on  $L_3$ 

$$\therefore P, Q, R \text{ are collinear, } \therefore \overrightarrow{PQ} || \overrightarrow{PR}$$

$$\Rightarrow \frac{\lambda}{\lambda - 1} = \frac{-\mu}{-1} = \frac{-1}{-v}$$

$$\Rightarrow \mu = \frac{\lambda}{\lambda - 1}, v = \frac{\lambda - 1}{\lambda}$$

Clearly from above that  $\lambda \neq 0, 1$ 

$$\therefore Q\left(0, \frac{\lambda}{\lambda - 1}, 1\right)$$

a. For 
$$Q = \hat{k} - \frac{1}{2}\hat{j}$$
  $\frac{\lambda}{\lambda - 1} = -\frac{1}{2} \Rightarrow 3\lambda = +1$ , which is possible.

b. For 
$$Q = \hat{k}$$
  
 $\frac{\lambda}{\lambda - 1} = 0 \Rightarrow \lambda = 0$ , not possible

c. For 
$$Q = \hat{k} + \hat{j}$$
  
 $\frac{\lambda}{\lambda - 1} = 1 \Rightarrow \lambda = \lambda - 1$ , not possible

d. For 
$$Q = \hat{k} + \frac{1}{2}\hat{j}$$

$$\frac{\lambda}{\lambda - 1} = \frac{1}{2} \Rightarrow 2\lambda = \lambda - 1 \Rightarrow \lambda = -1, \text{ which is possible.}$$

Hence options (a) and (d) are correct and options (b) and (c) are incorrect.

38. (a) the equation of hyperbola is 
$$\frac{x^2}{9} + \frac{y^2}{16} = 1$$



**Explanation:** For the given ellipse  $\frac{x^2}{25} + \frac{y^2}{16} = 1 \Rightarrow e = \sqrt{1 - \frac{16}{25}} = \frac{3}{5}$ 

 $\Rightarrow$  Eccentricity of hyperbola =  $\frac{5}{3}$ 

Let the hyperbola be  $\frac{x^2}{A^2} + \frac{y^2}{B^2} = 1$  then

$$B^2 = A^2 \left( rac{25}{9} - 1 
ight) = rac{16}{9} A^2 \mathrel{\dot{.}} \mathrel{\dot{.}} rac{x^2}{4^2} - rac{9y^2}{16 A^2} = 1$$

As it passes through focus of ellipse i.e. (3, 0)

 $\therefore$  Equation of hyperbola is  $\frac{x^2}{9} + \frac{y^2}{16} = 1$ 

Its focus is (5, 0) and vertex is (3, 0).

39. (a) does NOT intersect  $y = (x + 3)^2$ 

(c) intersects y = x + 2 exactly at one point

**Explanation:**  $[(x+2)(x+2+y)]^{\frac{dy}{dy}} - y^2 = 0$ , Put y = (x+2)t

$$\Rightarrow \frac{dy}{dx} = (x+2)\frac{dt}{dx} + t$$

$$(x+2)^2 = 0$$
 or  $(1+t)((x+2)\frac{dt}{dx}+t) - t^2 = 0$ 

$$(x+2)(1+t)\frac{dt}{dx} + t = 0$$

$$\left(rac{1+t}{t}
ight)dt = -rac{dx}{x+2}$$

 $\ln t + t = -\ln (x + 2) + c$ 

$$\Rightarrow \ln\left(\frac{y}{x+2}\right) + \left(\frac{y}{x+2}\right)$$

$$\Rightarrow$$
 -ln (x + 2) + c

$$\ln y - \ln (x+2) + \frac{y}{x+2} = -\ln(x+2) + c$$

$$\ln y + \frac{y}{x+2} = c$$

$$\ln 3 + 1 = c \Rightarrow \ln y + \frac{y}{x+2} = \ln 3e$$

i. In y + 
$$\frac{y}{x+2}$$
 = In 3e = ln (x + 2) + 1 = ln 3 + 1

 $\Rightarrow$  one solution

ii. 
$$y = (x+3)^2 \Rightarrow \ln(x+3)^2 + \frac{(x+3)^2}{x+2} = \ln 3 + 1$$

$$2 \ln(x+3) + \frac{(x+2)^2 + 1 + 2(x+2)}{x+2} = \ln 3 + 1$$

$$g(x) = 2 \ln(x+3) + (x+2) + 2 + \frac{1}{(x+2)}$$
 - In 3 - 1

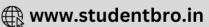
$$g(x) = \frac{2}{(x+3)} + 1 + 0 - \frac{1}{(x+2)^2} = \frac{2(x+2)^2 - (x+3)}{(x+3)(x+2)^2} + 1 > 0$$

Since x > 0 given and g(0) > 0, therefore g(x) will never intersect x-axis when x > 0.

- 40. (a) Probability that the chosen ball is green, given that the selected bag is B<sub>3</sub>, equals  $\frac{3}{8}$ 
  - (d) Probability that the chosen ball is green equals  $\frac{39}{80}$







$$\therefore P(B_1) = \frac{3}{10}, P(B_2) = \frac{3}{10}, P(B_3) = \frac{4}{10}$$

$$\mathrm{P}(rac{\mathrm{G}}{\mathrm{B}_1}) = rac{5}{10}, \mathrm{P}(rac{\mathrm{G}}{\mathrm{B}_2}) = rac{5}{8}, P\left(rac{G}{B_3}
ight) = rac{3}{8}$$

i. 
$$P(B_3 \cap G) = P(B_3) P\left(\frac{G}{B_3}\right) = \frac{4}{10} imes \frac{3}{8} = \frac{3}{20}$$

 $\therefore$  Probability that the selected bag is B<sub>3</sub> and the chosen ball is green equals  $\frac{3}{10}$  is not true.

ii. 
$$P(\frac{G}{B_3}) = \frac{3}{8}$$

 $\therefore$  Probability that the chosen ball is green, given that the selected bag is B<sub>3</sub>, equals  $\frac{3}{8}$  is true.

iii. : 
$$P(\frac{B_3}{G})$$

$$= \frac{P(G/B_3)P(B_3)}{P(G/B_1)P(B_1)+P(G/B_2)P(B_2)+P(G/B_3)P(B_3)}$$

$$= \frac{\frac{\frac{3}{8} \times \frac{4}{10}}{\frac{5}{10} \times \frac{3}{10} + \frac{5}{8} \times \frac{3}{10} + \frac{3}{8} \times \frac{4}{10}} = \frac{\frac{12}{80}}{\frac{15}{100} + \frac{15}{80} + \frac{12}{80}}$$

$$= \frac{12}{80} \times \frac{400}{60+75+60} = \frac{60}{195} = \frac{4}{13}$$

... Probability that the selected bag is B3, given that the chosen ball is green, equals  $\frac{5}{13}$  is not true.

iv. 
$$P(G) = P(\frac{G}{B_1})P(B_1) + P(\frac{G}{B_2})P(B_2) + P(\frac{G}{B_3})P(B_3)$$
  
 $= \frac{5}{10} \times \frac{3}{10} + \frac{5}{8} \times \frac{3}{10} + \frac{3}{8} \times \frac{4}{10}$   
 $= \frac{60+75+60}{400} = \frac{195}{400} = \frac{39}{80}$ 

 $\therefore$  Probability that the chosen ball is green equals  $\frac{39}{80}$  is true.

# 41.9

**Explanation:** 

Let 
$$M = \begin{bmatrix} a_1 & a_2 & a_3 \\ b_1 & b_2 & b_3 \\ c_1 & c_2 & c_3 \end{bmatrix}$$

$$\therefore M \begin{bmatrix} 0 \\ 1 \\ 0 \end{bmatrix} = \begin{bmatrix} -1 \\ 2 \\ 3 \end{bmatrix} \Rightarrow M \begin{bmatrix} 1 \\ -1 \\ 0 \end{bmatrix} = \begin{bmatrix} 1 \\ 1 \\ -1 \end{bmatrix}$$

$$M \cdot \begin{bmatrix} 1 \\ 1 \\ 1 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 12 \end{bmatrix}$$

$$\Rightarrow \begin{bmatrix} a_2 \\ b_2 \\ c_2 \end{bmatrix} = \begin{bmatrix} -1 \\ 2 \\ 3 \end{bmatrix}, \begin{bmatrix} a_1 - a_2 \\ b_1 - b_2 \\ c_1 - c_2 \end{bmatrix} = \begin{bmatrix} 1 \\ 1 \\ -1 \end{bmatrix}$$

$$\begin{bmatrix} a_1 + a_2 + a_3 \\ b_1 + b_2 + b_3 \end{bmatrix} = \begin{bmatrix} 0 \\ 0 \\ 12 \end{bmatrix}$$



$$\Rightarrow$$
 a<sub>2</sub> = -1, b<sub>2</sub> = 2, c<sub>2</sub> = 3, a<sub>1</sub> - a<sub>2</sub> = 1, b<sub>1</sub> - b<sub>2</sub> = 1, c<sub>1</sub> - c<sub>2</sub> = -1

$$\Rightarrow a_1 + a_2 + a_3 = 0, b_1 + b_2 + b_3 = 0, c_1 + c_2 + c_3 = 12$$

$$\therefore a_1 = 0, b_2 = 2, c_3 = 7$$

$$\Rightarrow$$
 Sum of diagonal elements =  $0 + 2 + 7 = 9$ 

42.4.0

**Explanation:** 

$$3x^2 + x - 1 = 4 \left| x^2 - 1 \right|$$

Case 1: If  $x \in [-1, 1]$ 

$$3x^2 + x - 1 = -4x^2 + 4$$

$$\Rightarrow 7x^2 + x - 5 = 0 : D = 141 > 0$$

: Equation has two roots

Case 2: If 
$$x \in (-\infty, -1] \cup [1, \infty)$$

$$3x^2 + x - 1 = 4x^2 - 4$$

$$\Rightarrow x^2 - x - 3 = 0$$
 : D = 13 > 0

: Equation has two roots So, total 4 roots

43.8

**Explanation:** 

$$\left((\log_2 9)^2\right)^{rac{1}{\log_2(\log_2 9)}} imes (\sqrt{7})^{rac{1}{\log_4 7}}$$

$$= (\log_2 9)^{2 imes \log_{(\log_2 9)} 2} imes 7^{rac{1}{2} imes \log_7 4}$$

$$= (\log_2 9)^{\log_{(\log_2 9)} 4} \times 7^{\log_7 2} = 4 \times 2 = 8$$

44. 1

Explanation:

Let 
$$g(x) = e^{f(x)}, \forall x \in R$$

$$\Rightarrow$$
 g'(x) = ef(x) - f'(x)

$$\Rightarrow$$
 f'(x) changes its sign from positive to negative in the neighbourhood of x = 2009

$$\Rightarrow$$
 f(x) has local maxima at x = 2009.

So, the number of local maximum is one.

45.3.0

**Explanation:** 

Given expansion 
$$\left(ax^2 + \frac{70}{27b \, b}\right)^4$$

$$T_{r+1} = {}^{4}C_{r}(ax^{2})^{4-r}(\frac{70}{27bx})^{r}$$

$$T_{r+1} = {}^{4}C_{r}(ax^{2})^{4-r}(\frac{70}{27bx})^{r}$$

$$= = {}^{4}C_{r}a^{4-r}(\frac{70}{27b})^{r} \cdot x^{8-3r}$$

Here, 
$$8 - 3r = 5 \Rightarrow r = 1$$



So, coefficient of 
$$x^5 = {}^4C_1a^3 \cdot \frac{70}{27 \text{ b}}$$

For expansion 
$$\left(ax - \frac{1}{bx^2}\right)^7$$

$$T_{r+1} = {}^{7}C_{r} (ax)^{7-r} \left(\frac{-1}{b^{2}}\right)^{r} = {}^{7}C_{r} a^{7-r} \left(\frac{-1}{b}\right)^{r} x^{7-3r}$$

Here, 
$$7 - 3r = -5 \Rightarrow r = 4$$

So, coefficient of 
$$x^{-5} = {}^{7}C_4 a^3 \left(\frac{-1}{b}\right)^3$$

A.T.Q, 
$${}^{4}C_{1}a^{3}\frac{70}{27b} = {}^{7}C_{4}a^{3} \cdot \frac{-1}{b^{3}}$$

$$\Rightarrow$$
 b =  $\frac{3}{2}$   $\Rightarrow$  2b = 3

Explanation:

Let locus point P(x, y).

: According to equation,

$$\left| rac{\sqrt{2}x+y-1}{\sqrt{3}} 
ight| \left| rac{\sqrt{2}x-y+1}{\sqrt{3}} 
ight| = \lambda^2$$

$$\Rightarrow \left| rac{2x^2 - (y-1)^2}{3} 
ight| = \lambda^2$$

So, C: 
$$|2x^2 - (y - 1)2| = \lambda^2$$

Let the line y = 2x + 1 meets C at two points R  $(x_1, y_1)$  and  $S(x_2, y_2)$ 

$$\Rightarrow$$
 y<sub>1</sub> = 2x<sub>1</sub> + 1 and y<sub>2</sub> = 2x<sub>2</sub> + 1 ...(i)

$$\Rightarrow$$
 (y<sub>1</sub> - y<sub>2</sub>) = 2(x<sub>1</sub> - x<sub>2</sub>)

$$\therefore RS = \sqrt{(x_1 - x_2)^2 + (y_1 - y_2)^2}$$

$$RS = \sqrt{5{(x_1 - x_2)}^2} = \sqrt{5} \, |x_1 - x_2|$$

On solving equations curve C and line y = 2x + 1, we get

$$|2x^2 - (2x)^2| = 3\lambda^2 \Rightarrow x^2 = \frac{3\lambda^2}{2}$$

$$\therefore RS = \sqrt{5} \left| \frac{2\sqrt{3}\lambda}{\sqrt{2}} \right| = \sqrt{30}\lambda = \sqrt{270} \Rightarrow 30\lambda^2 = 270 \Rightarrow \lambda^2 = 9$$

Explanation:

$$F(x) = \int_{0}^{x} f(t)dt$$

$$\Rightarrow F'(x) = f(x)$$

$$I = \int_{0}^{\pi} f'(x) \cdot \cos x \, dx + \int_{0}^{x} F(x) \cos x \, dx = 2 ...(i)$$

Using by parts

$$\Rightarrow$$
 I =  $(\cos x \cdot f(x))_0^{\pi} + \int_0^{\pi} \sin x \cdot f(x) dx + \int_0^{\pi} F(x) \cos x dx = 2$ 

$$\Rightarrow I = 6 - f(0) + \int_{0}^{\pi} \sin x \cdot F'(x) dx + \int_{0}^{\frac{\pi}{2}} F(x) \cos x dx = 2$$



$$I = 6 - f(0) + \int_{0}^{\pi} F'(x) \sin x dx + \int_{0}^{\pi} F(x) \cos x dx = 2$$

$$\Rightarrow I = 6 - f(0) + [\sin x F(x)]_{0}^{\pi} - \int_{0}^{\pi} F(x) \cos x dx + \int_{0}^{\pi} F(x) \cos x dx = 2$$

$$\Rightarrow I = 6 - f(0) + 0 = 2$$

$$\Rightarrow f(x) = 4$$

48.9

**Explanation:** 

Given,  $a_1 = 3$ , m = 5n and  $a_1, a_2, ...$ , is an AP.

$$\therefore \frac{S_m}{S_n} = \frac{S_{5n}}{S_n} \text{ is independent of n.}$$

$$= \frac{\frac{5n}{2}[2 \times 3 + (5n-1)d]}{\frac{n}{2}[2 \times 3 + (n-1)d]} = \frac{5\{(6-d) + 5n\}}{(6-d) + n}$$

If 
$$6 - d = 0 \Rightarrow d = 6$$

$$\therefore a_2 = a_1 + d = 3 + 6 = 9$$

or If d = 0, then  $\frac{S_m}{S_n}$  is independent of n.

$$\therefore a_2 = 9$$

